Claims:

- 21. (CURRENTLY AMENDED) A process for converting bulk nickel metal to nickel sulfate comprising the steps of:
- 1) providing the bulk nickel metal to a system having at least one enclosed reactor;
 - 2) supplying sulfuric acid solution to the system;
- supplying introducing oxygen into the systemsulfuric acid solution, said oxygen mixing with said sulfuric acid to form an oxygen containing sulfuric acid solution; and
- 4) reacting contacting said oxygen-containing and said sulfuric acid solution with the bulk nickel metal in the at least one enclosed reactor to produce nickel sulfate solution, wherein the exygen-containing-sulfuric acid solution is controlled maintained within a pH range; of about 0.1 to 6.0 wherein the difference between the maximum pH and the minimum pH of the pH runge is 5.9 or less.

22-45. (CANCELLED)

- 46 (CANCELLED)
- 47 (CURRENTLY AMENDED) The process of claim 21, wherein the pH range is 2.5 to 4.0 difference between the maximum pH and the minimum pH of the pH range is 2.5 or less.

k to exten 888

As the second se

- 48 (CANCELLED)
- 49. (PREVIOUSLY PRESENTED) The process of claim 21, wherein the pH range is 0.1 to 6.0.
- 50. (CANCELLED)
- 51. (CURRENTLY AMENDED) The process of claim 21, wherein pH range is of the oxygen containing sulfurio acid solution is controlled to a pH of less than 0.1 to 4.0.
- 52. (CURRENTLY AMENDED) The process of claim 21, wherein at least 1.8 kg of bulk nickel is converted to nickel sulfategreater than 1% of the bulk nickel is reacted per hour.
- 53. (CURRENTLY AMENDED) The process of claim 21, wherein the oxygeneontaining sulfuric acid solution reacts with bulk nickel-metal at a temperature of the at least one enclosed reactor is maintained at of 20 to 95 degrees Celsius-or-less.
- 54. (CURRENTLY AMENDED) The process of claim 53, wherein at least 1.8 kg of bulk nickel is converted to nickel sulfategreater than 1% of the bulk nickel is renoted per hour.

- 55. (CURRENTLY AMENDED) The process of claim 21, wherein oxygen gas is supplied to the system at a first pressure and sulfuric acid is supplied to the system at a second pressure.
- 56. (CURRENTLY AMENDED) The process of claim 2155, wherein the first pressure differs from the second pressure by an amount between 1 psi and 140 psi.
- 57. (CURRENTLY AMENDED) The process of claim 2455, wherein the first pressure is in a range of about 10 psi to about 149 psi and the second pressure is in a range of about 11 psi to about 150 psi.
- 58. (CURRENTLY AMENDED) The process of claim 21, wherein the bulk nickel metal has an average size greater than 0.1 mm.
- 59. (CURRENTLY AMENDED) The process of claim 21, wherein the bulk nickel metal has an average size of greater 0.254 cm in diameter.
- 60. (CURRENTLY AMENDED) The process of claim 21, wherein the bulk nickel metal has an average size of .245-0.254 cm to 3.81 cm in diameter.
- 61. (PREVIOUSLY PRESENTED) The process of claim 21, wherein the surface area of the bulk nickel is .20 square centimeters or greater.

06/15

PAGE

- 62. (PREVIOUSLY PRESENTED) The process of claim 21, wherein the bulk nickel metal comprises 99.99% nickel.
- 63. (CURRENTLY AMENDED) The process of claim 21, wherein said nickel sulfate solution has a nickel metal concentration of about 10 weight percenteentains a concentration of nickel of about 10 weight percent.
- 64. (PREVIOUSLY PRESENTED) The process of claim 21, wherein said step of supplying oxygen to the system comprises supplying air to the system.
- 65. (CURRENTLY AMENDED) A process for converting bulk nickel metal to nickel sulfate comprising the steps of:
- 1) providing the bulk nickel metal to a system having at least one enclosed reactor;
 - 2) supplying sulfuric acid solution to the system;
- 3) supplying introducing oxygen into the systemsulfuric acid solution, said oxygen mixing with said sulfuric acid to form an oxygen-containing sulfuric acid solution; and
- 4. reacting contacting said oxygen-containing and said sulfuric acid solution with the bulk nickel metal in the at least one enclosed reactor wherein the at least one enclosed reactor is maintained at a temperature of 20 degrees Celsius to 95 degrees Celsius to produce nickel sulfate solution, wherein the temperature during the reaction is less than 95 degrees Celsius.

reacted per hour.

- 66. (CURRENTLY AMENDED) The process of claim 4165, wherein at least 1,8 kg of bulk nickel is converted to nickel sulfatewherein greater than 1% of the bulk nickel is
- 67. (CURRENTLY AMENDED) The process of claim 4165, wherein the bulk nickel metal has an average size greater than 0.1 mm in diameter.
- 68 (CURRENTLY AMENDED) The process of claim 4165, wherein the bulk nickel metal has an average size of greater 0.254 cm in diameter.

- 69. (CURRENTLY AMENDED) A continuous process for converting bulk nickel metal to nickel sulfate comprising the steps of:
- 1.) providing bulk nickel metal to a system having at least one enclosed reactor;
 - 2) supplying sulfuric acid solution to the system;
- 3) supplying introducing oxygen into the system sulfuric acid solution, said oxygen mixing with said sulfuric acid to form an oxygen containing sulfuric acid solution; and
- 4.) reacting contacting the oxygen-containing and the sulfuric acid solution with the bulk nickel metal in the at lease one enclosed reactor, wherein the sulfuric acid solution is maintained within a pH range of about 0.1 to 4.0 and the at least one enclosed reactor is maintained at a temperature of 20 degrees Celsius to 100 degrees Celsius and wherein the controlled pH, wherein the temperature is less than 95 degrees Celsius and wherein the oxygen containing sulfuric acid is a controlled within a pH range—the difference between the maximum pH and the minimum pH of the controlled pH range being 2.5 or less.